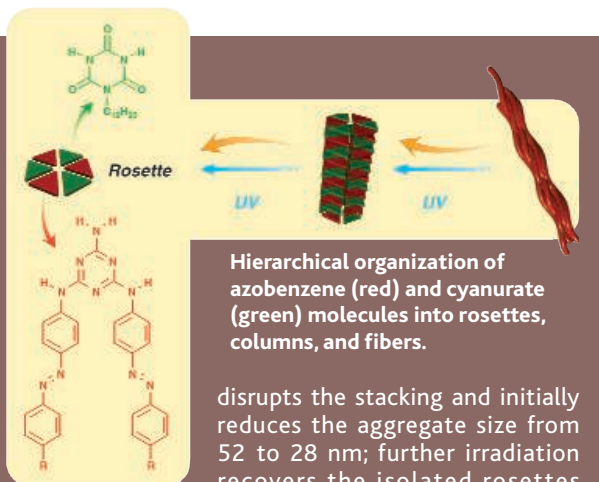


edited by Gilbert Chin

## CHEMISTRY

### Lightly Switched Gel

The formation of supramolecular assemblies can be controlled through light-induced structural movements, such as cis-trans isomerization, that alter the interactions between weakly bonding molecules. Yagai *et al.* have characterized disc-shaped hydrogen-bonded hexamers (rosettes) formed from two molecules: one a melamine bearing two long side chains containing azo groups and the other a much smaller cyanurate. In cyclohexane solution, the rosettes formed from the *trans*-azobenzene isomer can stack through aromatic interactions and bunch into columns that eventually intertwine and gel. Irradiation of the gel with ultraviolet light



**Hierarchical organization of azobenzene (red) and cyanurate (green) molecules into rosettes, columns, and fibers.**

disrupts the stacking and initially reduces the aggregate size from 52 to 28 nm; further irradiation recovers the isolated rosettes (8-nm aggregates). The dissociation is reversible, and exposure to visible light and subsequent storage in the dark yields the gel with total conversion of the cis isomers back to *trans*-azobenzenes. — PDS

*J. Am. Chem. Soc.* 10.1021/ja052645a (2005).

## BIOCHEMISTRY

### An On-Off Cycle

The mechanisms by which the activities of regulatory enzymes are themselves regulated range from tight-binding inhibitors to covalent modification. Sivaramakrishnan *et al.* have used a small molecule model in order to explore the chemical feasibility of regulating protein tyrosine phosphatase 1B (PTP1B) by reversible oxidation of its catalytic sulfhydryl. Structural analysis of inhibited PTP1B revealed the presence of a 3-isothiazolidinone adduct, in which the side chain of the active site cysteine had become covalently linked to the amide nitrogen of the next residue. Using a benzene scaffold to juxtapose a  $\beta$ -sulfinyl propionic acid ester and a monosubstituted amide nitrogen, they find that the in situ-generated sulfenic acid (RS-OH) is sufficiently reactive for the heterocycle to form under mild conditions (pH 7.5 and 37°C). In terms of how the corresponding biochemistry occurs, hydrogen peroxide oxidizes the sulfhydryl to the sulfenic acid, and glutathione opens the ring, forming a mixed disulfide that regenerates the free sulfhydryl. These reactions together would then serve as a redox cycle, switching phosphatase activity on and off. — GJC

*J. Am. Chem. Soc.* 10.1021/ja052599e (2005).

## IMMUNOLOGY

### Sweet Relations

Although bacteria are often thought of as harmful, it is now recognized that the many bacteria species harbored by our intestines are essential for our well-being. Aside from their roles in eliminating toxins and extracting nutrients, there is much interest in understanding

## PHYSICS

### Cold Atom Coupling

The ability to control the interaction strength between atoms within strongly interacting Fermi gases by sweeping a magnetic field across a Feshbach resonance provides a powerful experimental system in which to study many-body physics. One example is the crossover from a Bose-Einstein condensate (BEC) regime, in which the atoms are strongly coupled into pairs, to the weak-coupling regime that mimics Bardeen-Cooper-Schrieffer (BCS) coupling of electrons in superconducting metals. Although behavior on either side of the resonance is fairly well understood, of immediate interest is to find out what happens in the BEC-BCS crossover regime. However, determining the relative contributions of atom pairing mechanisms is an experimental and theoretical challenge.

Partridge *et al.* use a molecular spectroscopy technique to probe how the atoms pair up near the resonance. A laser is used to dress pairs of atoms

and project them onto a known molecular energy level. Locking the excitation rate onto the molecular level allows them to make a precise measurement of the contribution of each pairing mechanism. The technique should prove useful for closer studies of the many-body physics involved in these cold atom systems. — ISO

*Phys. Rev. Lett.* 95, 020404 (2005).

## CHEMISTRY

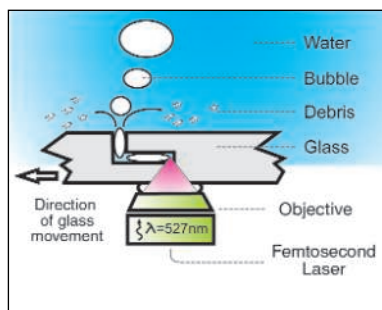
### A Bit of Bubbly

The popularity of the rapidly advancing field of microfluidics is due in part to the simplicity of making parts from polymers through etching or patterning methods. Some of the limitations of the

commonly used polydimethylsiloxane are solvent swelling, protein adsorption, leaching, and the inability to contain high pressures. Silica glass is often the best material for vessels for analytical and synthetic chemistry, but patterning glass at submicrometer dimensions is a challenge.

Ke *et al.* show that by using low-energy laser pulses, and by immersing the glass in a liquid, they can fabricate small channels in three dimensions. The laser is focused to a spot at the liquid/glass interface, so that a pulse both forms a hole in the glass and causes the liquid to expand as a bubble that pushes away the debris. Because the pulses are of low energy, the bubbles expand slowly and persist for much longer times than those associated with supersonic bubble collapse. The authors fabricated a number of architectures and channel designs, including a crisscross design that enhances the mixing of the fluids. — MSL

*Anal. Chem.* 10.1021/ac0505167 (2005).



**Schematic of the fabrication setup.**

